## **Asymmetric Synthesis of Clerodane Diterpenoids: Total Synthesis of (-)-Methyl Kolavenate**

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The asymmetric synthesis of (8R,9S,10R)-4,8,9-trimethyl-9-vinyl- $\Delta^{4}$ -3-octalone, a versatile intermediate for the syntheses of both *trans*- and *cis-*neo-clerodane diterpenoids, has been achieved by extension of Ender's asymmetric alkylation, and its utility is exemplified by the total synthesis of  $(-)$ -methyl kolavenate, the first example of a clerodane diterpenoid in optically active form.

The problem of the absolute configuration in clerodane diterpenoids was settled only recently by the reappraisal of physical methods,<sup>1</sup> while a chiral synthesis of a clerodane diterpenoid, which should corroborate the conclusion, has not been reported. We report here a method for the asymmetric synthesis of the  $\Delta^4$ -3-octalone intermediates (1), which are useful for the syntheses of both *trans-* and cis-clerodane diterpenoids,<sup>2</sup> and the total synthesis of  $(-)$ -methyl kolavenate utilizing this method.

Since the octalone intermediates (1) are derivable from **3,4-dimethylcyclohex-2-enone (4) ,2** our task amounts to the chiral preparation of the latter. We envisaged that **(4)** could be prepared from 6-methylcyclohex-2-enone **(3)** , which, in turn, should be obtainable in both enantiomeric forms from cyclohex-Zenone by the application of Ender's asymmetric alkylation.3 The methylation of cyclohexenone (S)-amino-2- (methoxymethy1)pyrrolidine (SAMP)-hydrazone **(2)** with methyl iodide followed by acidic hydrolysis was reported to give  $(3)$  in optical yields of  $61-\frac{75}{6}$  enantiomeric excess (e.e.). Improvements of this method were first studied, and we found that the use of methyl toluene-p-sulphonate as the alkylating reagent dramatically increased the optical purity of the product. Thus the reaction of the lithiated hydrazone with methyl toluene-p-sulphonate in tetrahydrofuran (THF) initially at  $-95^{\circ}$ C and then at up to  $-20^{\circ}$ C afforded the methylated product, which exhibits a 1H n.m.r. methyl signal only at  $\delta$  1.06 corresponding to one diastereoisomer. After quaternization (MeI) and hydrolysis (2 M HCl, pentane), **(3)**  with  $\lceil \alpha \rceil_D$  +91° (c 1.1, MeOH)<sup>4</sup> was obtained. The <sup>1</sup>H n.m.r. spectrum, measured in the presence of tris[(3-trifluoromethylhydroxymethylene)-(+)-camphorato]europium( $m$ ), showed no methyl doublet corresponding to the enantiomer, confirming the optical purity of **(3).** Treatment of **(3)** with methyl-lithium followed by oxidation with pyridinium chlorochromate (PCC) afforded (4),  $[\alpha]_D$  +95° (c 1.3, CHCl<sub>3</sub>); lit.<sup>5</sup>  $[\alpha]_{D}$  + 93° (CHCl<sub>3</sub>). *(R)*-3,4-Dimethyl-2-cyclohexenone (4), thus in hand, was converted according to the reported procedure<sup>2</sup> to yield the (10R)-octalone intermediate (1a),  $[\alpha]_D$  $+16^{\circ}$  (c 1.1, MeOH).

Starting with  $(R)$ -(1a), we synthesised  $(-)$ -methyl kolavenate, which was chemically correlated to a number of neoclerodane diterpenes, both *trans* and cis.6 After preliminary



Scheme 1. *Reagents:* i, Pr<sup>1</sup><sub>2</sub>NLi, THF, then p-MeC<sub>6</sub>H<sub>4</sub>SO<sub>3</sub>Me; ii, MeI, then 2 $M$  HCl, pentane; iii, MeLi, Et<sub>2</sub>O; iv, PCC, CH<sub>2</sub>Cl<sub>2</sub>; v,  $CH_2=CHMgBr·(\bar{B}u_{3}PCuI)_{4}$ , THF, then  $CH_2O$ ; vi, MeSO<sub>2</sub>Cl, Et<sub>3</sub>N,  $CH_2Cl_2$ ; vii, EtCOCH<sub>2</sub>CO<sub>2</sub>Me, NaOMe, benzene; viii, 2 M HCl, MeOH, reflux.





Scheme 2. Reagents: i, Et<sub>2</sub>AlCN, benzene-toluene; ii, L-Selectride, THF; iii,  $Me<sub>2</sub>CHCMe<sub>2</sub>BH<sub>2</sub>$ , THF, then  $H<sub>2</sub>O<sub>2</sub>$ , NaOH; iv, Bu<sup>t</sup>COCl, pyridine, then  $POCl<sub>3</sub>$ ; v, Bu<sup>i</sup><sub>2</sub>AlH, PhMe; vi, AcOH, H<sub>2</sub>O; vii,  $NH<sub>2</sub>NH<sub>2</sub>$ , KOH, HOCH<sub>2</sub>CH<sub>2</sub>OH; viii, MeSO<sub>2</sub>Cl, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>; ix, LiBr, THF; x, LiC=CH·EDA, Me<sub>2</sub>SO; xi, ZrCl<sub>2</sub>( $\eta$ -C<sub>5</sub>H<sub>5</sub>)<sub>2</sub>, Me<sub>3</sub>Al, then  $CICO<sub>2</sub>Me.$ 

investigations on the racemic material to find a more efficient synthetic path from **(la)** to **(9),** we performed the synthesis as in Scheme 2. Hydrocyanation of **(la)** followed by reduction with L-Selectride gave the 3 $\alpha$ -alcohol (5),  $[\alpha]_D$  +8° *(c* 1.1, MeOH) in 83% yield, which was hydroborated by thexylborane  $(Me<sub>2</sub>CHCMe<sub>2</sub>BH<sub>2</sub>)$  and oxidized. After selective protection of the primary hydroxy group in the diol obtained, the secondary OH was eliminated to produce (6),  $[\alpha]_D$  -126° *(c* 1.25, MeOH). The angular cyano group in *(6)* was transformed via an aldehyde to a methyl group and subsequent modification of the side chain functionality gave the bromide (7),  $[\alpha]_D - 41^\circ$  (c 0.35, MeOH). The addition of the extra four skeletal carbon units in **(9)** was carried out in two steps.? Firstly, treatment of **(7)** with lithium acetylide-ethylene-diamine (EDA) complex gave the acetylenic compound **(8)**,  $[\alpha]_D$  –46° (c 0.35, MeOH) in 71% yield with unavoidable formation of the elimination product  $(<25\%)$ . The final conversion was effected by the application of Negishi's zirconium-catalysed carboalumination method.' The metallation of **(8)** followed by treatment with methyl chloroformate furnished stereospecifically (-)-methyl kolavenate (9),  $[\alpha]_D$ -58" *(c* 0.28, MeOH), [lit.8 **-60.8"** (c 2.07, EtOH)] in 31% yield  $(49\%$  in the case of racemic material), $\ddagger$  accompanied by variable amounts of the hydromethylated product. Our synthetic product **(9)** was indistinguishable from an authentic sample by comparison of spectral data (i.r. and <sup>1</sup>H n.m.r.) and t.l.c. behaviour.

Starting from **(R)-amino-2-(methoxymethyl)pyrrolidine**  (RAMP)-hydrazone corresponding to **(2),** the method des-

**<sup>4</sup>**Hydrolysis of **(9)** to kolavenic acid was achieved by treatment with lithium hydroxide in aq. 1,2-dimethoxyethane at room temperature. cribed above can be used for the preparation of the (10s) octalone intermediate, which will be useful for the chiral synthesis of ent-neo-clerodane diterpenes. This fact as well as the availability of **(3)4** or **(4)5** also from natural chiral sources will make our method of asymmetric synthesis widely applicable.

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<sup>-</sup>f This conversion could also be achieved by the sequence: i, MeC $\equiv$ CCH<sub>2</sub>OMe, Bu<sup>n</sup>Li; ii, (CO<sub>2</sub>H)<sub>2</sub>; iii, MnO<sub>2</sub>, NaCN, AcOH, MeOH; but was less efficient.