## Asymmetric Synthesis of Clerodane Diterpenoids: Total Synthesis of (—)-Methyl Kolavenate

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The asymmetric synthesis of (8R,9S,10R)-4,8,9-trimethyl-9-vinyl- $\Delta^4$ -3-octalone, a versatile intermediate for the syntheses of both *trans*- and *cis*-neo-clerodane diterpenoids, has been achieved by extension of Ender's asymmetric alkylation, and its utility is exemplified by the total synthesis of (–)-methyl kolavenate, the first example of a clerodane diterpenoid in optically active form.

The problem of the absolute configuration in clerodane diterpenoids was settled only recently by the reappraisal of physical methods, which a chiral synthesis of a clerodane diterpenoid, which should corroborate the conclusion, has not been reported. We report here a method for the asymmetric synthesis of the  $\Delta^4$ -3-octalone intermediates (1), which are useful for the syntheses of both *trans*- and *cis*-clerodane diterpenoids, and the total synthesis of (-)-methyl kolavenate utilizing this method.

Since the octalone intermediates (1) are derivable from 3.4-dimethylcyclohex-2-enone (4),2 our task amounts to the chiral preparation of the latter. We envisaged that (4) could be prepared from 6-methylcyclohex-2-enone (3), which, in turn, should be obtainable in both enantiomeric forms from cyclohex-2-enone by the application of Ender's asymmetric alkylation.<sup>3</sup> The methylation of cyclohexenone (S)-amino-2-(methoxymethyl)pyrrolidine (SAMP)-hydrazone (2) with methyl iodide followed by acidic hydrolysis was reported to give (3) in optical yields of 61-75% enantiomeric excess (e.e.). Improvements of this method were first studied, and we found that the use of methyl toluene-p-sulphonate as the alkylating reagent dramatically increased the optical purity of the product. Thus the reaction of the lithiated hydrazone with methyl toluene-p-sulphonate in tetrahydrofuran (THF) initially at -95°C and then at up to -20°C afforded the

(2) (3)

(3)

(3)

(3)

(4)

(1a) R = Me

(1b) R = H

Scheme 1. Reagents: i,  $Pri_2NLi$ , THF, then  $p\text{-MeC}_6H_4SO_3Me$ ; ii, MeI, then  $2\,\text{M}$  HCl, pentane; iii, MeLi,  $Et_2O$ ; iv, PCC,  $CH_2Cl_2$ ; v,  $CH_2\text{-CHMgBr}\cdot(Bun_3PCuI)_4$ , THF, then  $CH_2O$ ; vi,  $MeSO_2CI$ ,  $Et_3N$ ,  $CH_2Cl_2$ ; vii,  $EtCOCH_2CO_2Me$ , NaOMe, benzene; viii,  $2\,\text{M}$  HCl, MeOH, reflux.

methylated product, which exhibits a <sup>1</sup>H n.m.r. methyl signal only at  $\delta$  1.06 corresponding to one diastereoisomer. After quaternization (MeI) and hydrolysis (2 M HCl, pentane), (3) with  $[\alpha]_D$  +91° (c 1.1, MeOH)<sup>4</sup> was obtained. The <sup>1</sup>H n.m.r. spectrum, measured in the presence of tris[(3-trifluoromethylhydroxymethylene)-(+)-camphorato]europium(III), showed no methyl doublet corresponding to the enantiomer, confirming the optical purity of (3). Treatment of (3) with methyl-lithium followed by oxidation with pyridinium chlorochromate (PCC) afforded (4),  $[\alpha]_D$  +95° (c 1.3, CHCl<sub>3</sub>); lit.<sup>5</sup>  $[\alpha]_D$  + 93° (CHCl<sub>3</sub>). (R)-3,4-Dimethyl-2-cyclohexenone (4), thus in hand, was converted according to the reported procedure<sup>2</sup> to yield the (10R)-octalone intermediate (1a),  $[\alpha]_D$  +16° (c 1.1, MeOH).

Starting with (R)-(1a), we synthesised (-)-methyl kolavenate, which was chemically correlated to a number of neoclerodane diterpenes, both *trans* and *cis*.<sup>6</sup> After preliminary

Scheme 2. Reagents: i,  $Et_2AlCN$ , benzene-toluene; ii, L-Selectride, THF; iii,  $Me_2CHCMe_2BH_2$ , THF, then  $H_2O_2$ , NaOH; iv,  $Bu^tCOCl$ , pyridine, then  $POCl_3$ ; v,  $Bu^t_2AlH$ , PhMe; vi, AcOH,  $H_2O$ ; vii,  $NH_2NH_2$ , KOH,  $HOCH_2CH_2OH$ ; viii,  $MeSO_2Cl$ ,  $Et_3N$ ,  $CH_2Cl_2$ ; ix, LiBr, THF; x,  $LiC\equiv CH\cdot EDA$ ,  $Me_2SO$ ; xi,  $ZrCl_2(\eta-C_5H_5)_2$ ,  $Me_3Al$ , then  $CICO_2Me$ .

investigations on the racemic material to find a more efficient synthetic path from (1a) to (9), we performed the synthesis as in Scheme 2. Hydrocyanation of (1a) followed by reduction with L-Selectride gave the  $3\alpha$ -alcohol (5),  $[\alpha]_D + 8^\circ$  (c 1.1, MeOH) in 83% yield, which was hydroborated by thexylborane (Me<sub>2</sub>CHCMe<sub>2</sub>BH<sub>2</sub>) and oxidized. After selective protection of the primary hydroxy group in the diol obtained, the secondary OH was eliminated to produce (6),  $[\alpha]_D$  -126° (c 1.25, MeOH). The angular cyano group in (6) was transformed via an aldehyde to a methyl group and subsequent modification of the side chain functionality gave the bromide (7),  $[\alpha]_D$  -41° (c 0.35, MeOH). The addition of the extra four skeletal carbon units in (9) was carried out in two steps.† Firstly, treatment of (7) with lithium acetylide-ethylene-diamine (EDA) complex gave the acetylenic compound (8),  $[\alpha]_D$  –46° (c 0.35, MeOH) in 71% yield with unavoidable formation of the elimination product (<25%). The final conversion was effected by the application of Negishi's zirconium-catalysed carboalumination method.<sup>7</sup> The metallation of (8) followed by treatment with methyl chloroformate furnished stereospecifically (-)-methyl kolavenate (9),  $[\alpha]_D$  -58° (c 0.28, MeOH), [lit.8 -60.8° (c 2.07, EtOH)] in 31% yield (49% in the case of racemic material),‡ accompanied by variable amounts of the hydromethylated product. Our synthetic product (9) was indistinguishable from an authentic sample by comparison of spectral data (i.r. and <sup>1</sup>H n.m.r.) and t.l.c. behaviour.

Starting from (R)-amino-2-(methoxymethyl)pyrrolidine (RAMP)-hydrazone corresponding to (2), the method des-

cribed above can be used for the preparation of the (10S)-octalone intermediate, which will be useful for the chiral synthesis of *ent*-neo-clerodane diterpenes. This fact as well as the availability of (3)<sup>4</sup> or (4)<sup>5</sup> also from natural chiral sources will make our method of asymmetric synthesis widely applicable.

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<sup>†</sup> This conversion could also be achieved by the sequence: i, MeC≡CCH<sub>2</sub>OMe, Bu<sup>n</sup>Li; ii, (CO<sub>2</sub>H)<sub>2</sub>; iii, MnO<sub>2</sub>, NaCN, AcOH, MeOH; but was less efficient.

<sup>‡</sup> Hydrolysis of (9) to kolavenic acid was achieved by treatment with lithium hydroxide in aq. 1,2-dimethoxyethane at room temperature.